## Supporting Information:

Scheme S1. Optimized geometries (B3LYP/6-311+G*) and some important geometrical parameters, (bond length ( $\mathrm{R}, \AA$ ) and bond angle ((A, degree)) of different molecules

| Molecules |  |  |
| :---: | :---: | :---: |
| Bond Distances $R(1,2)=1.87: R(2.3)=R(1,3)=1.57$ | $\left[\mathbf{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}$ |  |
| $R(1,2)=1.87, R(2,3)=R(1,3)=1.57$ $R(1,4)=\mathrm{R}(2,4)=1.82$ | Bond Distances $\mathrm{R}(1,2)=1.67 ; \mathrm{R}(2,3)=\mathrm{R}(1,3)=1.56$ | $\mathrm{R}(2,3)=\mathrm{R}(8,9)=1.57$ |
| $R(4,5)=R(4,6)=2.14$ | $\mathrm{R}(1,4)=\mathrm{R}(2,4)=1.89$ | $\mathrm{R}(3,4)=\mathrm{R}(7,8)=1.51$ |
| $R(4,7)=R(4,8)=2.15, R(4,9)=2.18$ | $\mathrm{R}(4,5)=1.91, \mathrm{R}(4,8)=1.98$ | $\mathrm{R}(1,4)=\mathrm{R}(6,7)=1.60$ |
| $\mathrm{R}(5,6)=1.43, \mathrm{R}(6,7)=\mathrm{R}(7,8)=1.42$ | $\mathrm{R}(5,6)=\mathrm{R}(7,8)=1.56$ | $\mathrm{R}(1,5)=\mathrm{R}(6,5)=1.78$ |
| $\mathrm{R}(8,9)=\mathrm{R}(9,5)=1.42$ | $\mathrm{R}(6,7)=1.52, \mathrm{R}(5,8)=1.77$ | $R(2,5)=R(9,5)=2.09$ |
| Bond Angles | Bond Angles | Bond Angles |
| $\mathrm{A}(1,2,3)=53.28 ; \quad \mathrm{A}(3,1,2)=53.34$ | $\mathrm{A}(1,2,3)=\mathrm{A}(3,1,2)=57.57 ; \quad \mathrm{A}(2,3,1)=64.86$ | $\mathrm{A}(1,2,3)=\mathrm{A}(6,9,8)=104.84$ $\mathrm{~A}(2,3)=72.46 . \quad \mathrm{A}(7,8,9)=72.53$ |
| $\mathrm{A}(2,3,1)=73.38$ $\mathrm{~A}(5,6,7)=107.88$. | $\mathrm{A}(1,4,2)=52.64 ; \quad \mathrm{A}(4,1,2)=\mathrm{A}(1,2,4)=63.68$ | $\mathrm{A}(2,3,4)=72.46 ; \quad \mathrm{A}(7,8,9)=72.53$ $\mathrm{A}(3,4,1)=\mathrm{A}(6,7,8)=117.81$ |
| $\mathrm{A}(5,6,7)=107.88 ; \mathrm{A}(6,7,8)=107.78$ $\mathrm{~A}(8,9,5)=107.72$ | $\mathrm{A}(5,6,7)=73.80$ | $\begin{aligned} & \mathrm{A}(3,4,1)=\mathrm{A}(6,7,8)=117.81 \\ & \mathrm{~A}(4,1,2)=64.85 ; \quad \mathrm{A}(9,6,7)=64.89 \end{aligned}$ |
| $A(8,9,5)=107.72$ $A(7,8,9)=108.36 ; \quad A(9,5,6)=108.18$ | $\mathrm{A}(6,7,8)=116.08 ; \quad \mathrm{A}(7,8,5)=67.03$ $\mathrm{~A}(8,5,6)=103.09 ; \quad \mathrm{A}(5,4,8)=54.14$ | $A(4,1,2)=64.85, A(9,6,7)=64.89$ $A(1,5,2)=A(6,5,9)=53.21$ |
| $\mathrm{A}(1,4,2)=62.18$ | $\mathrm{A}(5,8,4)=64.95 ; \quad \mathrm{A}(4,5,8)=60.91$ | $\mathrm{A}(1,2,5)=54.10 ; \quad \mathrm{A}(5,9,6)=54.13$ |
| $\mathrm{A}(8,4,7)=38.61 ; \quad \mathrm{A}(7,4,6)=38.68$ | $\mathrm{A}(5,6,8)=41.42 ; \quad \mathrm{A}(5,7,8)=62.10$ | $A(5,6,9)=70.75 ; \quad A(2,1,5)=70.68$ |
| $\mathrm{A}(6,4,5)=39.02$ | $\mathrm{A}(6,5,7)=52.22 ; \quad \mathrm{A}(6,8,7)=61.54$ | $\mathrm{A}(5,9,8)=52.56 ; \quad \mathrm{A}(4,2,3)=52.60$ |
| $\mathrm{A}(5,4,9)=38.32 ; \quad \mathrm{A}(9,4,8)=38.23$ |  |  |


|  |  |  |
| :---: | :---: | :---: |
|  | $\left[\mathrm{Al}_{4}-\mathrm{Fe}^{2}-\mathrm{Al}_{4}\right]^{2-}$ <br> Bond Distances $\begin{aligned} & \mathrm{R}(1,2)=\mathrm{R}(4,5)=2.62 \\ & \mathrm{R}(2,3)=2.64 ; \mathrm{R}(1,4)=1.56 \\ & \mathrm{R}(7,8)=\mathrm{R}(7,9)=2.26 ; \\ & \mathrm{R}(3,7)=2.51 ; \mathrm{R}(6,7)=2.49 \\ & \mathrm{R}(3,4)=\mathrm{R}(5,6)=\mathrm{R}(1,6)=\mathrm{R}(1,7)=2.65 \\ & \mathrm{R}(4,7)=\mathrm{R}(2,7)=\mathrm{R}(5,7)=2.65 \end{aligned}$ <br> Bond Angles <br> $\mathrm{A}(1,2,6)=33.31 ; \quad \mathrm{A}(1,6,2)=32.82$ <br> $\mathrm{A}(2,1,6)=113.86 ; \mathrm{A}(3,4,5)=114.49$ <br> $\mathrm{A}(4,3,5)=\mathrm{A}(5,3,4)=32.53 ; \mathrm{A}(6,2,3)=86.35$ <br> $\mathrm{A}(2,3,5)=93.67 ; \mathrm{A}(3,5,6)=86.12$ <br> $\mathrm{A}(2,6,5)=93.86 ; \mathrm{A}(6,5,7)=38.59$ <br> $\mathrm{A}(7,5,8)=39.81 ; \mathrm{A}(8,5,9)=40.26$ <br> $\mathrm{A}(9,5,10)=39.27$ |  |



Cp-Fe-Cp
Bond Distances
$\mathrm{R}(1,2)=\mathrm{R}(2,3)=\mathrm{R}(3,4)=\mathrm{R}(4,5)=\mathrm{R}(1,5)=1.42$
$R(7,8)=R(8,9)=R(9,10)=R(10,11)=R(7,11)=1.42$
$R(1,6)=R(2,6)=R(3,6)=R(4,6)=R(5,6)=2.08$
$R(6,7)=R(6,8)=R(6,9)=R(6,10)=R(6,11)=2.08$ Bond Angles
$A(1,2,3)=A(2,3,4)=A(3,4,5)=A(4,5,1)=A(5,1,2)=108$
$\mathrm{A}(7,8,9)=\mathrm{A}(8,9,10)=\mathrm{A}(9,10,11)=\mathrm{A}(10,11,7)=\mathrm{A}(11,7,8)=108$
$\mathrm{A}(4,6,3)=\mathrm{A}(3,6,2)=\mathrm{A}(2,6,1)=\mathrm{A}(1,6,5)=\mathrm{A}(5,6,4)=40.15$
$\mathrm{A}(10,6,9)=\mathrm{A}(9,6,8)=\mathrm{A}(8,6,7)=\mathrm{A}(7,6,11)=\mathrm{A}(11,6,10)=40.15$

Scheme S2. Optimized geometries (B3LYP/6-311+G*) and some important geometrical parameters, (bond length ( $\mathrm{R}, \AA$ ) and bond angle ((A, degree)) of different molecules




## $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.20$
$R(1,5)=R(2,5)=R(3,5)=R(4,5)=2.36$
$\mathrm{R}(6,5)=\mathrm{R}(7,5)=2.07$
$R(5,8)=R(5,9)=2.09 ; R(5,10)=2.08$
$\mathrm{R}(6,7)=\mathrm{R}(7,8)=\mathrm{R}(8,9)=\mathrm{R}(9,10)=(10,1)=1.43$

## Bond Angles

$\mathrm{A}(1,2,3)=\mathrm{A}(2,3,4)=89.99$
$\mathrm{A}(3,4,1)=\mathrm{A}(2,1,4)=90.00$
$\mathrm{A}(1,5,2)=55.54 ; \quad \mathrm{A}(2,5,3)=55.62$
$\mathrm{A}(3,5,4)=55.60 ; \quad \mathrm{A}(1,5,4)=55.52$
$\mathrm{A}(2,1,3)=\mathrm{A}(2,4,3)=45.00$
$\mathrm{A}(1,2,4)=\mathrm{A}(1,3,4)=44.99$
$\mathrm{A}(6,5,7)=40.44 ; \quad \mathrm{A}(7,5,8)=40.15$
$\mathrm{A}(8,5,9)=39.85 ; \quad \mathrm{A}(9,5,10)=39.95$
$\mathrm{A}(6,5,10)=40.31 ; \mathrm{A}(6,7,8)=107.95$
$\mathrm{A} 7,8,9)=108.03$
$\mathrm{A} 8,9,10)=108.11 \quad \mathrm{~A}(9,10,1)=108.00$

$\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$
Bond Distances
$R(1,2)=R(2,3)=R(3,4)=R(4,1)=1.40$ $R(1,5)=R(2,5)=R(3,5)=R(4,5)=1.98$ $R(6,5)=R(8,5)=2.07$, $R(9,5)=R(10,5)=2.08 ; R(7,5)=2.06$, $R(6,7)=R(7,8)=R(8,9)=R(9,10)=(10,1)=1.43$

## Bond Angles

$\mathrm{A}(1,2,3)=\mathrm{A}(1,4,3)=\mathrm{A}(2,1,4)=90.00$ $\mathrm{A}(2,3,4)=89.99$
$\mathrm{A}(1,5,2)=41.26 \quad \mathrm{~A}(2,5,3)=41.28$ $\mathrm{A}(3,5,4)=41.27 ; \quad \mathrm{A}(1,5,4)=41.24$, $\mathrm{A}(2,1,3)=\mathrm{A}(2,4,3)=45.00$
$\mathrm{A}(1,2,4)=\mathrm{A}(1,3,4)=44.99$
$\mathrm{A}(6,5,7)=40.42 ; \quad \mathrm{A}(7,5,8)=40.50$
$\mathrm{A}(8,5,9)=40.30$
$\mathrm{A}(9,5,10)=40.10 \quad \mathrm{~A}(6,5,10)=40.17$
$\mathrm{A}(6,7,8)=107.92$
$\mathrm{A} 7,8,9)=107.95 ; \quad \mathrm{A}(8,9,10)=108.05$
$\mathrm{A}(9,10,1)=108.07$

$\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$
Bond Distances
$R(1,5)=R(4,5)=2.43 ; R(1,2)=R(1,4)=2.22$
$R(2,3)=2.24 ; \quad R(3,4)=2.21$
$\mathrm{R}(2,5)=\mathrm{R}(3,5)=2.29$
$\mathrm{R}(6,7)=1.59 ; \quad \mathrm{R}(7,8)=1.52 ; \quad \mathrm{R}(8,9)=1.57$
$R(5,9)=1.98 ; R(5,6)=1.85 ; R(6,9)=1.75$
Bond Angles
$\mathrm{A}(6,5,9)=54.16 ; \quad \mathrm{A}(5,9,6)=59.06$
$\mathrm{A}(5,6,9)=66.77$
$\mathrm{A}(6,7,9)=61.60 ; \quad \mathrm{A}(7,8,9)=71.48$
$\mathrm{A}(7,6,9)=65.19 ; \quad \mathrm{A}(7,9,8)=52.81$
$\mathrm{A}(1,5,2)=55.99 ; \quad \mathrm{A}(2,5,3)=58.50$
$\mathrm{A}(3,5,4)=55.56$
$\mathrm{A}(1,5,4)=54.33$;
$\mathrm{A}(2,1,3)=45.35 ; \quad \mathrm{A}(2,4,3)=45.64$
$\mathrm{A}(1,2,4)=45.09 ; \quad \mathrm{A}(1,3,4)=45.81$
$\mathrm{A}(2,1,4)=\mathrm{A}(1,2,3)=89.81$
$\mathrm{A}(1,4,3)=90.74 ; \quad \mathrm{A}(2,3,4)=89.63$

Scheme S3. Optimized geometries (B3LYP/6-311+G*) and some important geometrical parameters, (bond length $\left(\mathrm{R}, \AA\right.$ ), and bond angle( $\left(\mathrm{A}\right.$, degree) ) of $\mathrm{Mg}_{4} \mathrm{Be}_{4}$ and different $\left[\mathrm{Mg}_{4} \mathrm{Be}_{4}-\right.$ M] metal clusters



$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Be}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.16$ $R(5,6)=R(7,8)=R(6,7)=R(8,5)=3.46$ $R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.47$ $R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.47$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=90.00$ $\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=90.00$ $\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=51.81$ $\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=51.81$ $\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$ $\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Mg}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.10$ $R(5,6)=R(7,8)=3.39 ; R(6,7)=R(8,5)=$ 3.26
$R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.58$
$R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.53$
Bond Angles
$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=115.11$
$\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=64.82$
$\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=48.55$
$\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=48.55$ $\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$ $\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ti}$

## Bond Distances

$\mathrm{R}(1,2)=\mathrm{R}(2,3)=\mathrm{R}(3,4)=\mathrm{R}(4,1)=2.07$
$R(5,6)=R(7,8)=R(6,7)=R(8,5)=2.96$ $R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.57$ $R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.57$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=90.00$
$\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=90.00$
$\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=47.53$
$A(3,7,2)=A(2,6,1)=47.53$
$\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$
$\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ca}$
Bond Distances
$\mathrm{R}(1,2)=\mathrm{R}(2,3)=\mathrm{R}(3,4)=\mathrm{R}(4,1)=2.09$
$R(5,6)=R(7,8)=3.55$
$R(6,7)=R(8,5)=3.27$
$R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.53$
$R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.52$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=114.65$
$\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=65.32$
$\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=48.97$
$\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=48.97$
$\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$
$\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Cr}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.07$
$R(5,6)=R(7,8)=3.19$
$R(6,7)=R(8,5)=3.43$
$\mathrm{R}(4,5)=\mathrm{R}(4,8)=\mathrm{R}(2,6)=\mathrm{R}(2,7)=2.56$ $R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.51$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=115.92$
$\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=63.82$
$A(1,5,4)=A(4,8,3)=48.16$
$\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=48.16$
$\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$ $A(8,7,6)=A(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Zn}$

## Bond Distances

$\mathrm{R}(1,2)=\mathrm{R}(2,3)=\mathrm{R}(3,4)=\mathrm{R}(4,1)=2.11$
$\mathrm{R}(5,6)=\mathrm{R}(7,8)=2.22 ; \mathrm{R}(6,7)=\mathrm{R}(8,5)=3.23$
$R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.62$
$\mathrm{R}(1,5)=\mathrm{R}(1,6)=\mathrm{R}(3,8)=\mathrm{R}(3,7)=2.54$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=114.06$ $\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=65.88$ $\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=48.23$ $\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=48.23$ $\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$ $\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{B}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.21$ $R(5,6)=R(7,8)=R(6,7)=R(8,5)=3.19$ $R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.48$ $R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.48$

## Bond Angles

$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=90.00$
$\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=90.00$ $\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=52.77$
$\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=52.77$
$\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$
$\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

$\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{C}$

## Bond Distances

$R(1,2)=R(2,3)=R(3,4)=R(4,1)=2.19$ $R(5,6)=R(7,8)=R(6,7)=R(8,5)=3.05$ $R(4,5)=R(4,8)=R(2,6)=R(2,7)=2.50$ $R(1,5)=R(1,6)=R(3,8)=R(3,7)=2.50$

Bond Angles
$\mathrm{A}(1,4,3)=\mathrm{A}(1,2,3)=90.00$ $\mathrm{A}(4,1,2)=\mathrm{A}(2,3,4)=90.00$ $\mathrm{A}(1,5,4)=\mathrm{A}(4,8,3)=51.92$ $\mathrm{A}(3,7,2)=\mathrm{A}(2,6,1)=51.92$ $\mathrm{A}(6,5,8)=\mathrm{A}(5,8,7)=90.00$ $\mathrm{A}(8,7,6)=\mathrm{A}(7,6,5)=90.00$

Scheme S4. Optimized geometries (B3LYP/6-311+G*) and some important bond lengths(R, $\AA$ ) of two different $\mathrm{Be}_{6}-\mathrm{Mg}$ isomers and the transition state separating them.

|  |  |
| :---: | :---: |
| $8(\mathrm{~Bq})=$ Dummy atom at the center of the planar $\mathrm{Be}_{6}$ ring. |  |
| $\begin{aligned} & \mathrm{Be}_{6}-\mathrm{M} \\ & \mathbf{E}=\mathbf{- 2 8} \\ & \mathrm{R} 17=2.74, \mathrm{R} 27=2.71, \mathrm{R} 37=2.6 \end{aligned}$ | ( $\mathrm{C}_{2}$ ) TS <br> 364 (a.u.) $\mathrm{R} 47=2.74, \mathrm{R} 57=2.71, \mathrm{R} 67=2.63$ |

Table S1. Total Energy (E, au), electronegativity ( $\chi, \mathrm{eV}$ ), hardness ( $\eta, \mathrm{eV}$ ), electrophilicity ( $\omega, \mathrm{eV}$ ) of different molecules involved in the substitution reactions

| Molecules | $\mathbf{E}(\mathbf{a u})$ | $\boldsymbol{\chi}(\mathbf{e V})$ | $\boldsymbol{\eta}(\mathbf{e V})$ | $\boldsymbol{\omega}(\mathbf{e V})$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Al}_{4}{ }^{2-}$ | -969.741 | -3.629 | 1.957 | 3.364 |
| $\mathrm{~B}_{4}{ }^{2-}$ | -99.088 | -5.135 | 1.711 | 7.705 |
| $\mathrm{~B}_{3}{ }^{-}$ | -74.397 | -0.678 | 3.368 | 0.068 |
| $\mathrm{Cp}^{-}$ | -193.570 | -1.512 | 3.423 | 0.334 |
| $\mathrm{Cp-Fe}-\mathrm{Cp}$ | -1650.883 | 3.400 | 3.695 | 1.564 |
| $\left.[\mathrm{Cp-}-\mathrm{Fe}-\mathrm{Al}]_{4}\right]^{-}$ | -2427.117 | 0.066 | 1.747 | 0.001 |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{-}$ | -3203.326 | -2.391 | 1.423 | 2.008 |
| $\left[\mathrm{Cp-Fe}-\mathrm{B}_{4}\right]^{-}$ | -1556.565 | -0.301 | 2.128 | 0.021 |
| $\left[\mathrm{~B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}$ | -1462.183 | -3.183 | 1.512 | 3.351 |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}$ | -2332.783 | -2.961 | 1.649 | 2.658 |
| $\mathrm{~B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}$ | -1412.411 | 5.543 | 2.491 | 6.168 |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}$ | -2307.909 | 0.797 | 1.273 | 0.250 |
| $\left[\mathrm{~B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}$ | -1437.380 | 0.803 | 1.724 | 0.187 |
| ${\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}}^{2}$ | -1531.634 | 4.610 | 2.349 | 4.523 |

Table S2. Total Energy ( $\mathrm{E}, \mathrm{au}$ ), electronegativity ( $\chi, \mathrm{eV}$ ), hardness ( $\eta$, eV), electrophilicity ( $\omega, \mathrm{eV}$ ) of different molecules involved in the substitution reactions

| Molecules | $\mathbf{E ~ ( a u )}$ | $\boldsymbol{\chi}(\mathbf{e V})$ | $\boldsymbol{\eta}(\mathbf{e V})$ | $\boldsymbol{\omega}(\mathbf{e V})$ |
| :---: | :---: | :---: | :---: | :---: |
| $\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}$ | -1701.573 | -3.551 | 1.996 | 3.158 |
| $\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ | -3994.880 | -2.959 | 2.013 | 2.175 |
| $\left[\mathrm{~N}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$ | -2848.227 | -3.121 | 1.923 | 2.532 |
| $\left[\mathrm{~B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ | -2703.703 | 0.675 | 1.950 | 0.117 |
| $\left[\mathrm{~B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}$ | -1462.183 | -3.183 | 1.512 | 3.351 |
| $\left[\mathrm{~B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$ | -1557.068 | 0.985 | 2.041 | 0.236 |
| $\mathrm{~B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}$ | -1412.411 | 5.543 | 2.491 | 6.168 |
| $\left[\mathrm{~B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}$ | -1581.862 | -3.332 | 1.272 | 4.365 |
| $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ | -2822.930 | 0.019 | 2.589 | 0.000 |
| $\left[\mathrm{~B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$ | -2728.518 | -3.144 | 1.419 | 3.484 |
| $\left[{\left.\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}}^{-1676.292}\right.$ | 0.020 | 2.860 | 0.000 |  |
| $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}$ | -1650.883 | 3.400 | 3.695 | 1.564 |
| $\mathrm{Cp}^{-}$ | -193.570 | -1.512 | 3.423 | 0.334 |
| $\mathrm{~B}_{3}{ }^{-}$ | -74.397 | -0.678 | 3.368 | 0.068 |
| $\mathrm{~B}_{4}{ }^{2-}$ | -99.088 | -5.135 | 1.711 | 7.705 |
| $\mathrm{~N}_{4}{ }^{2-}$ | -218.781 | -7.633 | 3.836 | 7.595 |
| $\mathrm{P}_{4}{ }^{2-}$ | -1365.475 | -4.560 | 2.514 | 4.135 |

Table S3. Point group (PG), Nucleus independent chemical shift (NICS, ppm) values at different rings of molecules

| Molecules | PG | NICS(0),ppm |  |
| :---: | :---: | :---: | :---: |
| $\mathrm{Al}_{4}{ }^{2-}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | -34.473 |  |
| $\mathrm{~B}_{4}{ }^{2-}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | -44.785 |  |
| $\mathrm{~B}_{3}{ }^{-}$ | $\mathrm{D}_{3 \mathrm{~h}}$ | -73.599 |  |
| $\mathrm{Cp}^{-}$ | $\mathrm{D}_{5 \mathrm{~h}}$ | -12.531 |  |
| $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}$ | $\mathrm{D}_{5 \mathrm{~h}}$ | $\mathrm{Cp}(-45.871)$ | $\mathrm{Cp}(-45.871)$ |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Cp}\right]^{-}$ | $\mathrm{C}_{\mathrm{s}}$ | $\mathrm{Al}_{4}(-19.329)$ | $\mathrm{Cp}(-31.470)$ |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}$ | $\mathrm{C}_{2 \mathrm{~h}}$ | $\mathrm{Al}_{4}(2,3,5,6)$ | $(29583.330)$ |
| $\left[{\left.\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}}^{\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}}\right.$ | $\mathrm{C}_{1}$ | $\mathrm{~B}_{4}(-16.178)$ | $\mathrm{Cp}(-56.558)$ |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}$ | $\mathrm{C}_{2}$ | $\mathrm{~B}_{4}(-25.549)$ | $\mathrm{B}_{4}(-25.680)$ |
| $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}$ | $\mathrm{C}_{\mathrm{s}}$ | $\mathrm{Al}_{4}(158.627)$ | $\mathrm{B}_{4}(-30.502)$ |
| $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}$ | $\mathrm{D}_{2 \mathrm{~d}}$ | $\mathrm{~B}_{3}(-31.871)$ | $\mathrm{B}_{3}(-31.867)$ |
| $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}$ | $\mathrm{C}_{\mathrm{s}}$ | $\mathrm{B}_{3}(-2.813)$ | $\mathrm{Al}_{4}(-10.268)$ |
| ${\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}}$ | $\mathrm{C}_{\mathrm{s}}$ | $\mathrm{B}_{3}(9.598)$ | $\mathrm{B}_{4}(9.530)$ |

Table S4. Point group (PG), Nucleus independent chemical shift (NICS, ppm) values at different rings of molecules

| Molecules | PG | NICS(0),ppm |  |
| :---: | :---: | :---: | :---: |
| $\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}$ | $\mathrm{S}_{8}$ | $\mathrm{N}_{4}(-32.859)$ | $\mathrm{N}_{4}(-32.896)$ |
| $\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$ | $\mathrm{S}_{8}$ | $\mathrm{P}_{4}(-8.988)$ | $\mathrm{P}_{4}(-9.005)$ |
| $\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$ | $\mathrm{C}_{4 \mathrm{v}}$ | $\mathrm{N}_{4}(-20.631)$ | $\mathrm{P}_{4}(-19.478)$ |
| $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ | $\mathrm{C}_{2 \mathrm{v}}$ | $\mathrm{P}_{4}(24.924)$ | $\mathrm{B}_{3}(5.191)$ |
| $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}$ | $\mathrm{C}_{2}$ | $\mathrm{B}_{4}(-24.490)$ | $\mathrm{B}_{4}(-8.252)$ |
| $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$ | $\mathrm{C}_{\text {s }}$ | $\mathrm{N}_{4}(-59.576)$ | $\mathrm{B}_{3}(-61.762)$ |
| $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}$ | $\mathrm{D}_{2 \mathrm{~d}}$ | $\mathrm{B}_{3}(-31.871)$ | $\mathrm{B}_{3}(-31.867)$ |
| $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}$ | $\mathrm{C}_{1}$ | $\mathrm{N}_{4}(-7.598)$ | $\mathrm{B}_{4}(-4.076)$ |
| [ $\left.\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ | $\mathrm{C}_{\mathrm{s}}$ | $\mathrm{P}_{4}(-15.059)$ | Cp(-39.158) |
| $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}$ | $\mathrm{C}_{1}$ | $\mathrm{B}_{4}(-24.450)$ | $\mathrm{P}_{4}(-8.252)$ |
| $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$ | $\mathrm{C}_{1}$ | Cp(-50.737) | $\mathrm{N}_{4}(-28.157)$ |
| Cp-Fe-Cp | $\mathrm{D}_{5 \mathrm{~h}}$ | Сp (-45.871) | Cp (-45.871) |
| $\mathrm{Cp}^{-}$ | $\mathrm{D}_{5 \mathrm{~h}}$ | -12.531 |  |
| $\mathrm{B}_{3}{ }^{-}$ | $\mathrm{D}_{3 \mathrm{~h}}$ | -73.599 |  |
| $\mathrm{B}_{4}{ }^{\text {- }}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | -44.785 |  |
| $\mathrm{N}_{4}{ }^{2-}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | 4.051 |  |
| $\mathrm{P}_{4}{ }^{\text {- }}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | 9.692 |  |

Table S5: Atomic charges ( $Q_{k,,}(\mathrm{NPA})$ ) and Fukui functions $\left(f_{k}{ }^{+}, f_{k}{ }^{-} \mathrm{eV}\right.$, (NPA)) values for different molecules involved in the substitution reactions

| Molecules | Unit | Atomic charge ( $Q_{\boldsymbol{k}}$ ) | $\mathrm{f}_{\boldsymbol{k}}{ }^{+}$(NPA) | $f_{k}{ }^{-}$(NPA) |
| :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} \mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp} \\ \left(\mathrm{D}_{5 \mathrm{~h}}\right) \end{gathered}$ | $\begin{aligned} & \mathrm{Cp} \\ & \mathrm{Cp} \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & \hline-0.251,-0.251,-0.251,-0.251,-0.251 \\ & -0.251,-0.251,-0.251,-0.251,-0.251 \end{aligned}$ $0.218$ | $\begin{gathered} \hline-0.074,0.034,0.005,0.005,0.034 \\ 0.005,0.003,-0.074,0.034,0.005 \\ 0.899 \end{gathered}$ | $0.037,0.038,0.036,0.036,0.038$ $0.036,0.038,0.037,0.038,0.036$ $0.286$ |
| $\begin{gathered} {\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{-}} \\ \left(\mathrm{C}_{\mathrm{s}}\right) \end{gathered}$ | $C p$ <br> $\mathrm{Al}_{4}$ <br> Fe | $\begin{gathered} \hline-0.339,-0.340,-0.343,-0.345,-0.347 \\ 0.261,0.253,0.250,0.254 \\ -1.408 \end{gathered}$ | $\begin{gathered} \hline 0.002,0.002,0.005,0.005,0.008 \\ 0.333,0.331,0.325,0.331 \\ -0.371 \end{gathered}$ | $\begin{gathered} \hline 0.021,0.036,0.009,0.004,0.020 \\ 0.186,0.077,0231,0.074 \\ 0.280 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{2 \mathrm{~h}}\right) \end{gathered}$ | $\mathrm{Al}_{4}$ <br> $\mathrm{Al}_{4}$ <br> Fe | $\begin{gathered} -0.204,-0.196,-0.038,0.336 \\ -0.042,0.336,-0.203,-0.194 \\ -1.797 \end{gathered}$ | $\begin{gathered} 0.206,0.215,0.089,0.147 \\ 0.089,0.148,0.209,0.214 \\ -0.316 \end{gathered}$ | $\begin{gathered} \hline 0.236,0.194,0.050,0.076 \\ 0.052,0.074,0.237,0.191 \\ -0.110 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}} \\ \left(\mathrm{C}_{1}\right) \end{gathered}$ | Cp <br> $B_{4}$ <br> Fe | $\begin{gathered} -0.300,-0.281,-0.305,-0.256,-0.336 \\ -0.320,-0.136,-0.234,-0.029 \\ -0.029 \end{gathered}$ | $\begin{gathered} 0.003,0.027,0.025,0.019,0.010 \\ 0.042,0.068,0.027,0.028 \\ 0.576 \end{gathered}$ | $\begin{gathered} 0.012,0.055,0.025,-0.002,0.052 \\ 0.114,0.239,-0.048,0.043 \\ 0.547 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{2}\right) \end{gathered}$ | $\mathrm{B}_{4}$ <br> $B_{4}$ <br> Fe | $\begin{gathered} 0.065,-0.298,-0.376,-0.286 \\ -0.376,-0.286,0.065,-0.299 \\ -0.209 \end{gathered}$ | $\begin{gathered} 0.010,0.001,0.018,0.031 \\ 0.018,0.030,0.010,0.001 \\ 0.881 \end{gathered}$ | $\begin{gathered} 0.006,-0.030,0.070,0.248 \\ 0.072,0.243,0.005,-0.042 \\ 0.427 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{Al}_{4}-\mathrm{Fe}_{-} \mathrm{B}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{\mathrm{s}}\right) \end{gathered}$ | $\mathrm{Al}_{4}$ <br> $B_{4}$ <br> Fe | $\begin{gathered} 0.291,0.025,0.287,0.121 \\ -0.364,-0.278,-0.178,-0.267 \\ -1.638 \end{gathered}$ | $\begin{gathered} 0.263,0.121,0.283,0.111 \\ 0.023,-0.028,0.160,0.065 \\ 0.002 \end{gathered}$ | $\begin{gathered} 0.107,0.282,0.079,0.287 \\ 0.023,0.035,0.122,0.062 \\ 0.002 \end{gathered}$ |
| $\begin{gathered} \mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3} \\ \left(\mathrm{D}_{2 \mathrm{~d}}\right) \end{gathered}$ | $B_{3}$ <br> $B_{3}$ <br> Fe | $\begin{gathered} 0.093,0.092,-0.047 \\ 0.093,-0.047,0.093 \\ -0.278 \end{gathered}$ | $\begin{gathered} 0.115,0.110,0.284 \\ 0.112,0.287,0.118 \\ -0.026 \end{gathered}$ | $\begin{gathered} 0.014,0.015,0.204 \\ 0.014,0.204,0.014 \\ 0.536 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}} \\ \left(\mathrm{C}_{\mathrm{s}}\right) \end{gathered}$ | $\mathrm{Al}_{4}$ $B_{3}$ <br> Fe | $\begin{gathered} 0.268,-0.025,-0.316,0.026 \\ -0.124,-0.163,-0.125 \\ -0.540 \end{gathered}$ | $\begin{gathered} 0.282,0.213,0.082,0.308 \\ 0.159,0.159,0.159 \\ -0.363 \end{gathered}$ | $\begin{gathered} -0.012,-0.070,0.244,0.252 \\ -0.038,0.176,-0.041 \\ 0.489 \end{gathered}$ |
| $\begin{gathered} {\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}} \\ \left(\mathrm{C}_{\mathrm{s}}\right) \end{gathered}$ | $\begin{aligned} & \mathrm{B}_{4} \\ & \mathrm{~B}_{3} \\ & \mathrm{Fe} \end{aligned}$ | $\begin{gathered} -0.065,-0.258,-0.011,-0.189 \\ -0.169,-0.166,-0.168 \\ 0.026 \\ \hline \end{gathered}$ | $\begin{gathered} \hline 0.254,0.073,0.167,0.095 \\ 0.146,0.210,0.146 \\ -0.091 \\ \hline \end{gathered}$ | $\begin{gathered} \hline 0.187,0.146,0.002,-0.016 \\ 0.028,0.315,0.028 \end{gathered}$ <br> 0.311 |
| $\begin{gathered} \mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3} \\ \left(\mathrm{C}_{1}\right) \end{gathered}$ | Cp <br> $B_{3}$ <br> Fe | $\begin{gathered} \hline-0.272,-0.248,-0.272,-0.256,-0.269 \\ 0.031,-0.016,0.026 \\ 0.102 \end{gathered}$ | $\begin{gathered} 0.037,0.044,0.032,0.041,0.038 \\ 0.208,0.294,0.206 \\ -0.002 \end{gathered}$ | $\begin{gathered} \hline 0.018,0.057,0.011,0.048,0.033 \\ 0.002,0.315,0.003 \\ 0.371 \\ \hline \end{gathered}$ |

Table S6: Atomic charges ( $Q_{k}$, NPA) and Fukui functions $\left(f_{k}{ }^{+}, f_{k}^{-}, \mathrm{eV}\right.$, (NPA)) values for nucleophilic and electrophilic attacks respectively for different molecules involved in the substitution reactions

| Molecules | Unit | Atomic charges ( $Q_{k}$ ) | $\mathrm{f}^{+}$(NPA) | $\boldsymbol{f}_{\boldsymbol{k}}{ }^{-}$(NPA) |
| :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} {\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}} \\ \left(\mathrm{S}_{8}\right) \end{gathered}$ | $\mathrm{N}_{4}$ | -0.321,-0.323,-0.321,-0.320 | 0.013,0.011,0.013,0.007 | 0.091,0.086,0.090,0.086 |
|  | $\mathrm{N}_{4}$ | -0.321,-0.320,-0.322,-0.323 | 0.013,0.007,0.013, 0.011 | 0.088,0.089,0.088,0.089 |
|  | Fe | 0.570 | 0.913 | 0.294 |
| $\begin{gathered} {\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}} \\ \left(\mathrm{S}_{8}\right) \end{gathered}$ | $\mathrm{P}_{4}$ | -0.138,-0.138,-0.138,-0.138 | 0.029,0.029,0.029,0.029 | 0.093,0.094,0.095,0.094 |
|  | $\mathrm{P}_{4}$ | -0.138,-0.138,-0.138,-0.138 | 0.029,0.029,0.029,0.029 | 0.095,0.094,0.093,0.094 |
|  | Fe | -0.897 | 0.771 | 0.247 |
| $\begin{gathered} {\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{4} \mathrm{~V}\right) \end{gathered}$ | $\mathrm{N}_{4}$ | -0.259,-0.259,-0.259,-0.259 | 0.011,0.010,0.011,0.010 | 0.053,0.052,0.053,0.052 |
|  | $\mathrm{P}_{4}$ | -0.247,-0.247,-0.247,-0.247 | 0.029,0.029,0.028,0.030 | $0.148,0.146,0.148,0.146$ |
|  | Fe | 0.024 | 0.843 | 0.200 |
| $\begin{gathered} {\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}} \\ \left(\mathrm{C}_{2} \mathrm{~V}\right) \end{gathered}$ | $\mathrm{P}_{4}$ | -0.059,-0.152,-0.057 | 0.124.284,0.121 | -0.029,0.298,-0.031 |
|  | $\mathrm{B}_{3}$ | -0.117,-0.117,-0.117,-0.117 | 0.163,0.163,0.151,0.154 | 0.110,0.115,0.112,0.114 |
|  | Fe | -0.262 | -0.161 | 0.311 |
| $\begin{gathered} {\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{2}\right) \end{gathered}$ | $\mathrm{B}_{4}$ | $0.065,-0.298,-0.376,-0.286$ | 0.010,0.001,0.018,0.031 | 0.006,-0.030,0.070,0.248 |
|  | $\mathrm{B}_{4}$ | -0.376,-0.286,0.065,-0.267 | 0.018,0.030,0.010,0.001 | 0.072,0.243, $0.005,-0.042$ |
|  | Fe | -0.209 | 0.881 | 0.427 |
| $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$ <br> ( $\mathrm{C}_{\mathrm{s}}$ ) | $\mathrm{B}_{3}$ | -0.357,-0.106,-0.146 | 0.091,0.374,0.170 | 0.201,0.274,0.056 |
|  | $\mathrm{N}_{4}$ | -0.180,-0.252,-0.257,-0.181 | 0.074,0.100,0.095, 0.074 | $0.045,0.061,0.063,0.052$ |
|  | Fe | 0.478 | 0.024 | 0.249 |
| $\begin{gathered} \mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3} \\ \left(\mathrm{D}_{2} \mathrm{~d}\right) \end{gathered}$ | ${ }^{B_{3}}$ | 0.093, 0.092,-0.047 | $0.115,0.110, .284$ | 0.014,0.015,0.204 |
|  | $\mathrm{B}_{3}$ | 0.093,-0.047,0.093 | 0.112,0.287,0.118 | 0.014,0.204, 0.014 |
|  | Fe | -0.278 | -0.026 | 0.536 |
| $\begin{gathered} {\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{1}\right) \end{gathered}$ | $\mathrm{B}_{4}$ | -0.266,-0.144,-0.402,-0.322 | 0.071,0.059,0.052,0.077 | -0.049,0.078,0.104,0.298 |
|  | $\mathrm{N}_{4}$ | -0.279,-0.286,-0.325,-0.288 | 0.017,0.024,0.016,0.0191 | 0.029,0.070,0.042,0.098 |
|  | Fe | 0.312 | 0.656 | 0.329 |
| $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}$ <br> (C $\mathrm{C}_{\mathrm{s}}$ ) | ${ }^{\text {Cp }}$ | -0.265,-0.269,-0.269,-0.267,-0.263 | 0.045,0.012,0.012,0.045,-0.087 | 0.020,0.019,0.019,0.020,0.020 |
|  | $\mathrm{P}_{4}$ | -0.129,-0.127,-0.131,-0.132 | 0.032,0.030,0.033,0.032 | $0.144,0.145,0.144,0.144$ |
|  | Fe | -. 230 | 0.824 | 0.189 |
| $\begin{gathered} {\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}} \\ \left(\mathrm{C}_{1}\right) \end{gathered}$ | $\mathrm{B}_{4}$ | -0.178,0.029,-0.391,-0.256 | 0.021,0.108,0.012,0.096 | $-0.070,-0.011,0.095,0.249$ |
|  | $\mathrm{P}_{4}$ | -0.140,-0.324,-0.315,-0.114 | 0.045,-0.007,0.007,0.050 | 0.086,0.127,0.115,0.066 |
|  | Fe | -0.311 | 0.660 | 0.342 |
| $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}$$\left(\mathrm{C}_{1}\right)$ | Cp | -0.299,-0.298,-0.0297,-0.296,-0.295 | 0.032,0.033,-0.069,-0.069,0.058 | 0.038,0.038,0.039,0.040,0.041 |
|  | $\mathrm{N}_{4}$ | -0.248,-0.246,-0.248,-0.247 | 0.065,-0.064,-0.065,0.065 | 0.080,0.081,0.085, 0.079 |
|  | Fe | 0.432 | 0.928 | 0.302 |
| [Cp-Fe-Cp] | ${ }^{\text {Cp }}$ | -0.251,-0.251,-0.251,-0.251,-0.251 | -0.074,0.034,0.005,0.005,0.034 | ${ }^{0.037,0.038,0.036,0.036,0.038}$ |
|  | Cp | -0.251,-0.251,-0.251,-0.251,-0.251 | 0.005,0.003,-0.074,0.034,0.005 | 0.036,0.038,0.037,0.038, , 0.036 |
| ( $\mathrm{D}_{5 \mathrm{~h}}$ ) | Fe | 0.218 | 0.899 | 0.286 |

Table S7: Reaction enthalpy ( $\Delta \mathrm{H}, \mathrm{Kcal} / \mathrm{mole}$ ) and reaction electrophilicity ( $\Delta \omega, \mathrm{eV}$ ) values of different molecules which are involved in the substitution reactions

| No. | Reactions | $\begin{gathered} \Delta H \\ \text { (Kcal/mole) } \end{gathered}$ | $\Delta \omega(\mathrm{eV})$ |
| :---: | :---: | :---: | :---: |
| 1 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{+}+\mathrm{Cp}^{-}$ | -99.305 | -4.592 |
| 2 | $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{-}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -81.875 | -1.022 |
| 3 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}+\mathrm{Cp}{ }^{-}$ | -104.845 | -8.913 |
| 4 | $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -63.361 | -4.041 |
| 5 | $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}+\mathrm{B}_{3}{ }^{-}$ | -102.995 | -1.537 |
| 6 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{B}_{3}{ }^{-}$ | -69.878 | -4.473 |
| 7 | $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -153.169 | -9.213 |
| 8 | $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -173.312 | -13.618 |
| 9 | $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Cp}\right]^{-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -92.741 | -4.715 |
| 10 | $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}+\mathrm{B}_{3}{ }^{-}$ | -113.68 | -5.229 |
| 11 | $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{-}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -87.202 | -0.393 |
| 12 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}$ | -23.841 | 3.648 |
| 13 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{3}\right]^{-}+\mathrm{Al}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{B}_{3}{ }^{-}$ | -93.72 | -0.825 |
| 14 | $\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{Al}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{Al}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{Al}_{4}{ }^{2-}$ | -10.866 | -3.692 |
| 15 | $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{Cp}=\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{B}_{3}{ }^{-}$ | -29.251 | -1.911 |
| 16 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{Cp}^{-}=\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}+\mathrm{B}_{3}{ }^{-}$ | -45.733 | -3.225 |

Table S8: Reaction enthalpy $(\Delta \mathrm{H})$ and reaction electrophilicity $(\Delta \omega)$ values of different molecules which are involved in the substitution reactions

Table 9: Energy ( $\mathrm{E}, \mathrm{au}$ ), electronegativity ( $\chi, \mathrm{eV}$ ), hardness ( $\eta, \mathrm{eV}$ ), and electrophilicity ( $\omega, \mathrm{eV}$ ) for different atoms invole in the formation of different $\left[\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{M}\right]$ clusters

| No | Reactions | $\Delta \mathbf{H}$ (Kcal/mole) | $\Delta \omega(\mathrm{eV})$ |
| :---: | :---: | :---: | :---: |
| 1 | $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -171.014 | -13.458 |
| 2 | $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{B}_{3}{ }^{-}$ | -75.733 | -4.605 |
| 3 | $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{P}_{4}{ }^{2-}$ | -36.930 | -3.339 |
| 4 | $\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{P}_{4}{ }^{2-}=\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -134.084 | -10.118 |
| 5 | $\left[\mathrm{B}_{3}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{P}_{4}{ }^{2-}=\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{B}^{3-}$ | -61.692 | -2.010 |
| 6 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{P}_{4}{ }^{2-}$ | -23.917 | -2.578 |
| 7 | $\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{P}_{4}{ }^{2-}$ | -25.863 | -3.102 |
| 8 | $\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{P}_{4}{ }^{2-}$ | -25.108 | -2.833 |
| 9 | $[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P} 4]^{-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{P}_{4}{ }^{2-}$ | -34.942 | -3.459 |
| 10 | $\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{P}_{4}{ }^{2-}$ | -14.997 | -2.260 |
| 11 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{P}_{4}{ }^{2-}$ | -32.416 | -3.703 |
| 12 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -171.850 | -12.049 |
| 13 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}$ | -12.056 | -1.097 |
| 14 | $\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{B}_{4}{ }^{2-}=\left[\mathrm{B}_{4}-\mathrm{Fe}-\mathrm{B}_{4}\right]^{2-}+\mathrm{N}_{4}{ }^{2-}$ | -8.498 | -1.125 |
| 15 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}+\mathrm{P}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{Cp}^{-}$ | -91.175 | -5.365 |
| 16 | $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{P}_{4}{ }^{2-}=\left[\mathrm{P}_{4}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -29.618 | -1.627 |
| 17 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{B}_{3}+\mathrm{P}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{P}_{4}\right]^{-}+\mathrm{B}_{3}{ }^{-}$ | -136.908 | -8.590 |
| 18 | $\mathrm{Cp}-\mathrm{Fe}-\mathrm{Cp}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{Cp}^{-}$ | -126.118 | -8.824 |
| 19 | $\left[\mathrm{Cp}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{-}+\mathrm{N}_{4}{ }^{2-}=\left[\mathrm{N}_{4}-\mathrm{Fe}-\mathrm{N}_{4}\right]^{2-}+\mathrm{Cp}^{-}$ | -45.646 | -4.102 |


| Atoms | $\mathbf{E}(\mathbf{a u})$ | $\boldsymbol{\chi}(\mathbf{e V})$ | $\boldsymbol{\eta}(\mathbf{e V})$ | $\boldsymbol{\omega}(\mathbf{e V})$ |
| :---: | :---: | :---: | :---: | :---: |
| Li | -7.491 | 3.087 | 2.529 | 1.884 |
| Na | -162.287 | 3.003 | 2.418 | 1.864 |
| K | -599.926 | 2.510 | 1.986 | 1.586 |
| Be | -14.671 | 4.445 | 4.671 | 2.115 |
| Mg | -200.093 | 3.752 | 3.976 | 1.770 |
| Ca | -677.576 | 3.085 | 3.069 | 1.551 |
| Sc | -760.621 | 3.506 | 4.607 | 1.334 |
| Ti | -849.290 | 3.648 | 2.098 | 3.172 |
| Cr | -1044.224 | 3.879 | 1.560 | 4.821 |
| Zn | -1779.354 | 4.199 | 5.230 | 1.685 |
| C | -37.792 | 5.706 | 4.064 | 4.006 |
| B | -24.662 | 4.189 | 4.545 | 1.930 |

Table S10: Point group (PG), energy (E, au), electronegativity ( $\chi, \mathrm{eV}$ ), hardness ( $\eta, \mathrm{eV}$ ), and electrophilicity $(\omega, \mathrm{eV})$ for different metal clusters

| Molecules | PG | E (au) | $\boldsymbol{\chi}(\mathbf{e V})$ | $\boldsymbol{\eta}(\mathbf{e V})$ | $\boldsymbol{\omega}(\mathbf{e V})$ |
| :--- | :--- | :---: | :--- | :--- | :--- |
| $\mathrm{Be}_{4}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | -58.766 | 3.688 | 2.261 | 3.007 |
| $\mathrm{Mg}_{4}$ | $\mathrm{D}_{4 \mathrm{~h}}$ | -800.309 | 2.987 | 1.508 | 2.959 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -859.276 | 3.526 | 1.792 | 3.468 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Li}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -866.838 | 3.423 | 1.821 | 3.217 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Na}$ | $\mathrm{C}_{1}$ | -1021.618 | 3.301 | 1.792 | 3.040 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{K}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -1459.259 | 3.118 | 1.770 | 2.747 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Be}$ | $\mathrm{C}_{4 \mathrm{v}}$ | -874.050 | 3.694 | 2.020 | 3.377 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Mg}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -1059.404 | 3.545 | 1.874 | 3.353 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ca}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -1536.917 | 3.245 | 1.872 | 2.812 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Sc}$ | $\mathrm{C}_{4 \mathrm{v}}$ | -1620.007 | 3.319 | 1.805 | 3.051 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ti}$ | $\mathrm{C}_{4 \mathrm{v}}$ | -1708.722 | 3.580 | 1.562 | 4.103 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Cr}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -1903.673 | 2.906 | 1.131 | 3.734 |
| $\mathrm{Mg}_{4} \mathrm{Be} 4-\mathrm{Zn}$ | $\mathrm{C}_{2 \mathrm{v}}$ | -2638.652 | 3.766 | 1.881 | 3.770 |
| $\mathrm{Mg}_{4} \mathrm{Be} \mathrm{Be}_{4}-\mathrm{B}$ | $\mathrm{C}_{4 \mathrm{v}}$ | -884.168 | 3.401 | 1.954 | 3.752 |
| $\mathrm{Mg}_{4} \mathrm{Be} 4-\mathrm{C}$ | $\mathrm{C}_{4 \mathrm{v}}$ | -897.466 | 3.830 | 2.399 | 2.410 |

Table S11: Nucleus independent chemical shift (NICS(0), ppm) values at different rings of the molecules

| Molecules | $\left.\begin{array}{c}\text { NICS(0) } \\ \text { Ring(Mg }\end{array}\right)$ | $\left.\begin{array}{c}\text { NICS(0) } \\ \text { Ring(Be }\end{array}\right)$ |
| :--- | :---: | :---: | :--- | :---: | :---: |

Table S12: Formation reactions of different metal clusters and their reaction enthalpy ( $\Delta \mathrm{H}$, $\mathrm{Kcal} / \mathrm{mole}$ ) and reaction electrophilicity ( $\Delta \omega, \mathrm{eV}$ ) values

| Formation <br> Reactions | $\Delta \mathbf{H}$ <br> (Kcal/mole) | $\Delta \boldsymbol{\omega}$ <br> (eV) |
| :--- | :---: | :---: |
| $\mathrm{Mg}_{4}+\mathrm{Be}_{4}=\mathrm{Mg}_{4} \mathrm{Be}_{4}$ | -123.979 | -2.498 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Li}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Li}$ | -4.456 | -0.780 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Na}=\mathrm{Mg}_{4} \mathrm{Be} e_{4}-\mathrm{Na}$ | -34.067 | -0.743 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{K}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{K}$ | -35.452 | -0.556 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Be}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Be}$ | -23.163 | -2.206 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Mg}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Mg}$ | -21.830 | -1.885 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Ca}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ca}$ | -40.587 | -2.206 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Sc}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Sc}$ | -68.793 | -1.751 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Ti}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ti}$ | -97.014 | -2.537 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Cr}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Cr}$ | -108.320 | -4.555 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{Zn}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Zn}$ | -14.264 | -1.383 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{B}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{B}$ | -142.402 | -1.646 |
| $\mathrm{Mg}_{4} \mathrm{Be}_{4}+\mathrm{C}=\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{C}$ | -247.214 | -5.064 |

Table S13: Atomic charges $\left(Q_{k}(N P A)\right)$ and Fukui functions $\left(f_{k}^{+}, f_{k}^{-}, \mathrm{eV},(\mathrm{NPA})\right)$ and philicity $\left(\omega_{k}{ }^{+}\right.$, $\left.\omega_{k}{ }^{-}, \mathrm{eV},(\mathrm{NPA})\right)$ values for nucleophilic and electrophilic attacks respectively for different metal cluster

| Molecules | Unit | Atomic Charge $\left(Q_{k}\right)(N P A)$ | $f_{k}^{+}$( NPA) | $f_{k}^{\text {- }}$ ( NPA) | $\omega_{k}^{+}$(NPA) | $\omega_{k}{ }^{-}$(NPA) |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | $0.346,0.347$ | 0.136, 0.137 | 0.099, 0.099 | 0.472, 0.474 | 0.343, 0.342 |
|  |  | $0.346,0.346$ | 0.136, 0.137 | 0.099, 0.099 | 0.473, 0.474 | $0.343,0.342$ |
|  | $\mathrm{Be}_{4}$ | -0.211, -0.482 | 0.096, 0.130 | 0.156, 0.148 | 0.333, 0.451 | $0.541,0.513$ |
|  |  | -0.483, -0.210 | 0.130, 0.098 | 0.148, 0.153 | 0.452, 0.339 | 0.514, 0.530 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Li} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.333, 0.334 | 0.148, 0.148 | 0.154, 0.155 | 0.476, 0.476 | 0.497, 0.498 |
|  |  | 0.336, 0.336 | 0.148, 0.148 | 0.155, 0.154 | 0.477, 0.477 | 0.497, 0.498 |
|  | $\mathrm{Be}_{4}$ | -0.259, -0.564 | 0.049, 0.091 | 0.046, 0.100 | 0.158, 0.292 | 0.148, 0.321 |
|  |  | -0.575, -0.263 | 0.078, 0.047 | 0.091, 0.047 | 0.250, 0.153 | 0.294, 0.150 |
|  | Li | 0.322 | 0.143 | 0.097 | 0.459 | 0.314 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Na} \\ \left(\mathrm{C}_{1}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.339, 0.328 | 0.147, 0.145 | 0.145, 0.143 | 0.446, 0.440 | 0.441, 0.435 |
|  |  | 0.326, 0.315 | 0.138, 0.135 | 0.143, 0.140 | $0.419,0.410$ | $0.434,0.425$ |
|  | $\mathrm{Be}_{4}$ | -0.287, -0.569 | 0.039, 0.045 | 0.048, 0.067 | 0.117, 0.136 | 0.146, 0.203 |
|  |  | -0.491, -0.252 | 0.106, 0.044 | 0.112, 0.052 | $0.323,0.133$ | 0.341, 0.157 |
|  | Na | 0.291 | 0.203 | 0.151 | 0.616 | 0.458 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{K} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | $0.313,0.314$ | 0.136, 0.136 | 0.137, 0.137 | 0.374, 0.374 | 0.377, 0.375 |
|  |  | 0.313, 0.314 | 0.136, 0.136 | 0.137, 0.137 | 0.375, 0.374 | $0.378,0.375$ |
|  | $\mathrm{Be}_{4}$ | -0.302, -0.567 | 0.036, 0.063 | 0.058, 0.096 | $0.099,0.173$ | 0.159, 0.264 |
|  |  | -0.568, -0.298 | 0.063, 0.037 | 0.095, 0.052 | 0.173, 0.102 | 0.262, 0.143 |
|  | K | 0.482 | 0.256 | 0.150 | 0.704 | 0.413 |
| $\underset{\left(\mathrm{C}_{4 \mathrm{v}}\right)}{\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Be}}$ | $\mathrm{Mg}_{4}$ | 0.343, 0.344 | 0.106, 0.106 | 0.088, 0.088 | 0.358, 0.358 | 0.298, 0.298 |
|  |  | 0.344, 0.344 | 0.106, 0.106 | 0.088, 0.088 | 0.358, 0.359 | 0.298, 0.298 |
|  | $\mathrm{Be}_{4}$ | -0.400, -0.401 | 0.081, 0.081 | 0.139, 0.137 | 0.275, 0.274 | 0.470, 0.463 |
|  |  | -0.401, -0.401 | 0.081, 0.081 | 0.137, 0.139 | 0.274, 0.273 | 0.463, 0.468 |
|  | Be | 0.230 | 0.250 | 0.095 | 0.846 | 0.322 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Mg} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.266, 0.266 | 0.105, 0.105 | 0.163, 0.163 | 0.352, 0.352 | 0.547, 0.548 |
|  |  | 0.264, 0.264 | 0.104, 0.104 | 0.164, 0.164 | 0.348, 0.348 | 0.550, 0.551 |
|  | $\mathrm{Be}_{4}$ | -0.269, -0.489 | -0.020, 0.232 | 0.050, 0.078 | -0.050, 0.779 | 0.166, 0.261 |
|  |  | -0.491, -0.270 | 0.229, -0.020 | 0.080, 0.050 | 0.767, -0.050 | $0.267,0.167$ |
|  | Mg | 0.459 | 0.153 | 0.088 | 0.512 | 0.297 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ca} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.270, 0.270 | 0.090, 0.090 | 0.297, 0.297 | 0.254, 0.257 | 0.236, 0.234 |
|  |  | 0.270, 0.270 | 0.090, 0.091 | 0.297, 0.297 | 0.252, 0.254 | $0.235,0.234$ |
|  | $\mathrm{Be}_{4}$ | -0.311, -0.594 | -0.000, 0.172 | 0.205, -0.010 | -0.000, 0.483 | 0.576, -0.035 |
|  |  | -0.594, -0.310 | 0.170, 0.002 | -0.010, 0.202 | 0.477, 0.007 | -0.039, 0.569 |
|  | Ca | 0.729 | 0.295 | 0.286 | 0.830 | 0.803 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Sc} \\ \left(\mathrm{C}_{4 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.320, 0.319 | 0.082, 0.081 | 0.083, 0.083 | $0.249,0.247$ | 0.252, 0.253 |
|  |  | 0.320, 0.320 | 0.079, 0.079 | 0.082, 0.082 | $0.243,0.240$ | 0.251, 0.251 |
|  | $\mathrm{Be}_{4}$ | -0.093, -0.090 | 0.082, 0.081 | 0.081, 0.083 | 0.289, 0.289 | 0.247, 0.253 |
|  |  | -0.090, -0.093 | 0.079, 0.079 | 0.080, 0.082 | 0.271, 0.284 | 0.245, 0.251 |
|  | Sc | -0.913 | 0.082 | 0.344 | 0.940 | 1.048 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Ti} \\ \left(\mathrm{C}_{4 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.357, 0.357 | 0.128, 0.128 | 0.059, 0.059 | 0.527, 0.527 | 0.241, 0.241 |
|  |  | 0.357, 0.357 | 0.128, 0.128 | 0.059, 0.059 | 0.527, 0.527 | $0.241,0.241$ |
|  | $\mathrm{Be}_{4}$ | 0.068, 0.069 | 0.131, 0.132 | 0.107, 0.106 | 0.540, 0.541 | 0.440, 0.437 |
|  |  | 0.069, 0.068 | 0.131, 0.132 | 0.106, 0.107 | 0.541, 0.540 | 0.437, 0.440 |
|  | Ti | -1.703 | -0.040 | 0.337 | -0.160 | 1.384 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Cr} \\ \left(\mathrm{C}_{2 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.323, 0.323 | 0.122, 0.123 | 0.143, 0.143 | 0.455, 0.458 | 0.535, 0.534 |
|  |  | 0.323, 0.324 | 0.122, 0.123 | 0.143, 0.143 | 0.456, 0.459 | 0.535, 0.533 |
|  | $\mathrm{Be}_{4}$ | 0.036, -0.115 | 0.008, 0.019 | 0.021, 0.003 | 0.030, 0.071 | 0.079, 0.012 |
|  |  | -0.114, 0.038 | 0.019, 0.010 | 0.002, 0.091 | 0.072, 0.037 | 0.009, 0.071 |


|  | Cr | -1.139 | 0.454 | 0.382 | 1.696 | 1.426 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $\underset{\left(\mathrm{C}_{2 \mathrm{v}}\right)}{\mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{Zn}}$ | $\mathrm{Mg}_{4}$ | $0.254,0.254$ | 0.102, 0.099 | 0.162, 0.162 | 0.385, 0.375 | 0.609, 0.611 |
|  |  | 0.256, 0.255 | 0.104, 0.101 | 0.161, 0.161 | 0.393, 0.382 | 0.607, 0.609 |
|  | $\mathrm{Be}_{4}$ | -0.233, -0.395 | -0.011, 0.272 | 0.022, 0.134 | -0.040, 1.025 | 0.082, 0.505 |
|  |  | -0.391, -0.234 | 0.277, -0.020 | 0.128, 0.023 | 1.043, -0.080 | 0.483, 0.089 |
|  | Zn | 0.233 | 0.075 | 0.046 | 0.285 | 0.174 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{B} \\ \left(\mathrm{C}_{4 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.356, 0.355 | 0.956, 0.955 | 0.113, 0.115 | $0.359,0.355$ | 0.425, 0.433 |
|  |  | $0.355,0.356$ | 0.956, 0.955 | 0.113, 0.115 | $0.359,0.356$ | 0.425, 0.432 |
|  | $\mathrm{Be}_{4}$ | 0.489, 0.482 | 0.177,0.109 | 0.089, 0.094 | 0.438, 0.408 | $0.333,0.353$ |
|  |  | 0.482, 0.481 | 0.109, 0.109 | 0.094, 0.097 | 0.410, 0.409 | 0.351, 0.365 |
|  | B | -3.356 | 0.175 | 0.169 | 0.658 | 0.635 |
| $\begin{gathered} \mathrm{Mg}_{4} \mathrm{Be}_{4}-\mathrm{C} \\ \left(\mathrm{C}_{4 \mathrm{v}}\right) \end{gathered}$ | $\mathrm{Mg}_{4}$ | 0.351, 0.353 | 0.104, 0.092 | 0.094, 0.094 | 0.251, 0.221 | 0.226, 0.226 |
|  |  | 0.351, 0.352 | 0.090, 0.106 | 0.094, 0.094 | 0.218, 0.255 | $0.227,0.226$ |
|  | $\mathrm{Be}_{4}$ | $0.487,0.489$ | $0.034,0.251$ | $0.130,0.116$ | $0.082,0.605$ | $0.314,0.279$ |
|  |  | 0.487, 0.484 | 0.248, 0.033 | 0.122, 0.128 | 0.598, 0.079 | 0.294, 0.308 |
|  | C | -3.353 | 0.042 | 0.129 | 0.100 | 0.311 |

Table S14: Energy (E, au), Point Group (PG), Electronegativity ( $\chi, \mathrm{eV}$ ), Chemical Hardness ( $\eta$, eV ) and Electrophilicity ( $\omega, \mathrm{eV}$ ) values of two different $\mathrm{Be}_{6}-\mathrm{Mg}$ isomers.

| Isomers | $\mathbf{E ~ ( a . u . ) ~}$ | PG $^{2}$ | $\boldsymbol{\chi}(\mathbf{e V})$ | $\boldsymbol{\eta}(\mathbf{e V})$ | $\boldsymbol{\omega}(\mathbf{e V})$ |
| :--- | :---: | :---: | :---: | :---: | :---: |
| $\mathrm{Be}_{6}-\mathrm{Mg}(\mathrm{I})$ | -288.370 | $\mathrm{C}_{6 \mathrm{v}}$ | 3.715 | 1.949 | 3.541 |
| $\mathrm{Be}_{6}-\mathrm{Mg}(\mathrm{TS})$ | -288.364 | $\mathrm{C}_{2}$ | 3.848 | 1.952 | 3.792 |
| $\mathrm{Be}_{6}-\mathrm{Mg}$ (II) | -288.370 | $\mathrm{C}_{6 \mathrm{v}}$ | 3.713 | 1.949 | 3.537 |

Table S15: Nucleus independent chemical shift (NICS(0),ppm), atomic charges ( $\mathrm{Q}_{\mathrm{K}}$, NPA) and Fukui function ( $f k^{+}, f k, e \mathrm{eV},(\mathrm{NPA})$ ) values for nucleophilic and electrophilic attacks respectively of two different $\mathrm{Be}_{6}-\mathrm{Mg}$ isomers.

| Isomers | Unit | NICS(0) | Atomic Charges <br> $\mathbf{( \mathbf { Q } _ { \mathbf { K } } )}$ | $\boldsymbol{f k}^{+} \mathbf{( N P A )}$ | $\boldsymbol{f k}^{-}(\mathbf{N P A})$ |
| :---: | :--- | :---: | :---: | :---: | :---: |
| $\mathrm{Be}_{6}-\mathrm{Mg}(\mathrm{I})$ <br> $\left(\mathrm{C}_{6} \mathrm{~V}\right)$ | $\mathrm{Be}_{6}$ |  | $-0.042,-0.042,-0.042$ | $0.115,0.156,0.157$ | $0.155,0.155,0.155$ |
|  | Mg | 0.93 | $-0.042,-0.042,-0.042$ | $0.115,0.154,0.154$ | $0.155,0.155,0.155$ |
| $\mathrm{Be}_{6}-\mathrm{Mg}(\mathrm{II})$ | $\mathrm{Be}_{6}$ |  | 0.252 | 0.148 | 0.069 |
| $\left(\mathrm{C}_{6} \mathrm{~V}\right)$ | Mg | 1.00 | $-0.042,-0.041,-0.041$ | $0.155,0.1185,0.159$ | $0.156,0.1536,0.153$, |
|  |  | $0.042,-0.042$ | $0.156,0.1516,0.112$ | $0.154,0.157,0.157$ |  |

